

## The Reaction of 4-Methoxycoumarins with Prenyl Bromide: Synthesis of 4',4',5'-Trimethyl-dihydrofurano-coumarins and 2,2-Dimethyl-chromenopyrans

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The reaction of 7-hydroxy-4-methoxy- (**1a**), 7-hydroxy-4,5-dimethoxy- (**1b**) and 5-hydroxy-4,7-dimethoxy (**1c**)-2*H*-1-benzopyran-2-ones with prenyl bromide in acetone in the presence of anhydrous potassium carbonate gave the corresponding prenyloxycoumarins **1d**, **1e**, and **1f**. The prenyloxy coumarins **1d**, **1e**, **1f** on *Claisen* migration by refluxing in *N,N*-dimethyl aniline gave the corresponding 4',4',5'-trimethyl-dihydrofuranocoumarins **2a**, **2b**, and **3**. However, the reaction of **1a**, **1b**, and **1c** with 3-chloro-3-methyl-but-1-yne in acetone in presence of potassium carbonate and potassium iodide gave the corresponding propargyl ethers **1g**, **1h**, **1i**, which on refluxing in *N,N*-dimethyl aniline gave the corresponding 2,2-dimethylchromenopyrans **4a**, **4b**, and **5**. These can also be obtained directly if the reaction is carried out in the presence of dioxan.

(*Keywords*: 3-Chloro-3-methylbut-1-yne; *Claisen* migration; Propargyl ethers)

*Die Reaktion von 4-Methoxycumarinen mit Prenylbromid:  
Synthese von 4',4',5'-Trimethyl-dihydro-furanocumarinen und 2,2-Dimethyl-  
chromenopyranen*

Die Reaktion von 7-Hydroxy-4-methoxy- (**1a**), 7-Hydroxy-4,5-dimethoxy- (**1b**) und 5-Hydroxy-4,7-dimethoxy-2*H*-1-benzopyran-2-onen (**1c**) mit Prenylbromid in Aceton in der Gegenwart von wasserfreiem Kaliumcarbonat ergab die entsprechenden Prenylcoumarine **1d**, **1e** und **1f**. Diese ergaben durch *Claisen*-Wanderung bei Rückfluß in *N,N*-Dimethylanilin die entsprechenden 4',4',5'-Trimethyl-dihydrofuranocumarine **2a**, **2b** und **3**. Die Reaktion von **1a**, **1b** und **1c** mit 3-Chlor-3-methyl-1-butin in Aceton in Gegenwart von wasserfreiem Kaliumcarbonat und Kaliumjodid ergab die entsprechenden Propargylether **1g**, **1h** und **1i**, die ihrerseits bei Rückfluß in *N,N*-Dimethylanilin die jeweiligen 2,2-Dimethylchromenopyrane **4a**, **4b** und **5** ergaben. Letztere können auch direkt erhalten werden, wenn die Reaktion in Gegenwart von Dioxan durchgeführt wird.

### Introduction

The isoprenyl unit as 2,2-dimethylchromene ring or as 3,3-dimethyl allyl unit is a frequently occurring arrangement in natural products. The study of 4-hydroxycoumarin and its derivatives is of special importance in view of their being in tautomeric<sup>2,3</sup> equilibrium with the ketoform and its important physiological properties<sup>4,5</sup>. In view of this and due to the presence of coumarins with different modifications of isopentenyl group in nature, the reaction of 4-methoxycoumarins, viz., 7-hydroxy-4-methoxy-2*H*-1-benzopyran (**1a**), 7-hydroxy-4,5-dimethoxy-2*H*-1-benzopyran-2-one (**1b**) and 5-hydroxy-4,7-dimethoxy-2*H*-1-benzopyran-2-one (**1c**) with prenyl bromide and 3-chloro-3-methyl-but-1-yne has been carried out with a view to synthesize 4',4',5'-trimethyldihydrofuranocoumarins and 2,2-dimethylchromenopyrans.

### Results and Discussion

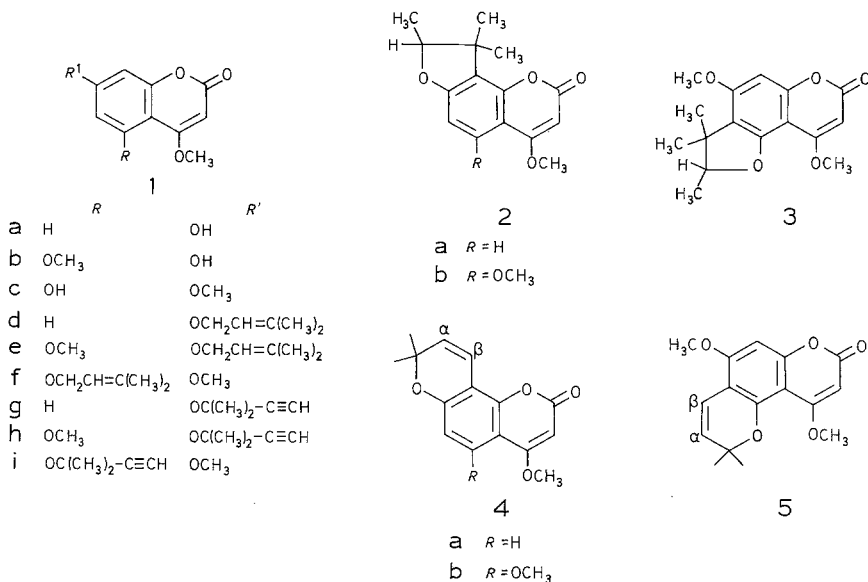
The reaction of 7-hydroxy-4-methoxy-2*H*-1-benzopyran-2-one (**1a**) with prenyl bromide in dry acetone in the presence of anhydrous potassium carbonate gave an alkali insoluble product **A**. Its <sup>1</sup>H-NMR spectrum showed besides other signals the presence of two singlets at  $\delta$  1.75 and 1.80 integrating for three protons each, which could be assigned to the gem. dimethyl group i.e.,  $-\text{CH}_2-\text{CH}=\text{C}(\text{CH}_3)_2$ ,  $\delta$  4.57 ( $J = 8$  Hz) integrating for two protons i.e. methylene protons and a multiplet at  $\delta$  5.54 integrating for one methyne proton. Hence, **A** was assigned the structure of 4-methoxy-7-prenyloxy-2*H*-1-benzopyran-2-one (**1d**). Similarly 7-hydroxy-4,5-dimethoxy-2*H*-1-benzopyran-2-one (**1b**) and 5-hydroxy-4,7-dimethoxy-2*H*-1-benzopyran-2-one (**1c**) gave 4,5-dimethoxy-7-prenyloxy-2*H*-1-benzopyran-2-one (**1e**) and 4,7-dimethoxy-5-prenyloxy-2*H*-1-benzopyran-2-one (**1f**). Similar results were obtained when the reaction was carried out in dioxan. **1d** on *Claisen* migration in boiling *N,N*-dimethylaniline gave a compound **2a** which was insoluble in alkali.

In the <sup>1</sup>H-NMR spectrum it showed the presence of three methyl groups, among which one appeared as a doublet at  $\delta$  1.32 ( $J = 7$  Hz) indicating it to be a secondary methyl group and the other two methyl groups as two singlets at  $\delta$  1.40 and 1.52. Further one hydrogen resonated as a quartet at 4.50 ( $J = 7$  Hz), which can be adjacent to ring oxygen atom. Finally two *ortho* coupled aromatic protons with  $J = 10$  Hz each appeared at  $\delta$  6.65 and 7.58.

Hence compound **2a** was assigned the structure of a 4-methoxy-4',4',5'-trimethyl-dihydrofurano(2',3',7,8)-2*H*-1-benzopyran-2-one.

However prenylation of 7-hydroxy-4-methoxy-2*H*-1-benzopyran-2-one (**1a**) with 3-chloro-3-methyl-but-1-yne in acetone in the presence of

anhydrous potassium carbonate and potassium iodide gave the propargyl ether **1g** as can be deduced from its <sup>1</sup>H-NMR spectrum which indicated the presence of two methyl groups as a singlet at  $\delta$  1.70 and acetylenic protons at  $\delta$  2.61 besides other signals. On refluxing with *N,N*-dimethylaniline **1g** gave 4-methoxy-8,8-dimethyl-2-oxo-2*H*,8*H*-



chromeno[2,3-*h*]-pyran (**4a**). Chromene **4a** could also be obtained in one step by carrying out the reaction with 3-chloro-3-methyl-but-1-yne in presence of potassium carbonate and potassium iodide in dioxan.

Identical results were obtained for 7-hydroxy-4,5-dimethoxy-2*H*-1-benzopyran-2-one (**1b**) and 5-hydroxy-4,7-dimethoxy-2*H*-1-benzopyran-2-one (**1c**). The structures of these compounds were in agreement with their spectral data which are presented in Table 1 and 2.

The coumarins **1b** and **1c**, required for the prenylation studies, have been obtained by catalytic debenzoylation of the earlier described<sup>3</sup> 7-benzyloxy-4,5-dimethoxy- and 5-benzyloxy-4,7-dimethoxy-2*H*-1-benzopyran-2-ones.

## Experimental

### *Reaction of Hydroxycoumarins in Acetone and Dioxan with Prenyl Bromide*

The appropriate hydroxycoumarin (**1a**, **1b** or **1c**) (**1g**) in dry acetone or dioxan (50 ml) was refluxed for 6 h. with prenyl bromide in presence of anhydrous K<sub>2</sub>CO<sub>3</sub>. The solution was filtered, the residue washed with more

Table 1. *Melting points and <sup>1</sup>H-NMR spectral data of compounds 1d-1i (in CDCl<sub>3</sub>)*

Com- pound	M.P. (°C)	—CH <sub>2</sub> —CH=C	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	—OCH <sub>3</sub> (s)	H-3	$\begin{matrix} \text{H}_3\text{C} & & \text{C} & & \text{CH} \\   & &   & &   \\ \text{C} & & \text{C} & & \text{C} \\   & &   & &   \\ \text{H}_3\text{C} & & \text{CH}_3 & & \text{H} \end{matrix}$	Aromatic Protons
<b>1d</b>	135-136	1.75 and 1.80 [2s, (CH <sub>3</sub> ) <sub>2</sub> ] 4.57 (d, —OCH <sub>2</sub> —, J = 8) 5.54 (m, —CH=, H-3)	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	3.95	5.54 (m, —CH=, H-3)	—	6.80 (m, H-6 and H-8) 7.65 (d, H-5, J = 10)
<b>1e</b>	138-139	1.75 and 1.80 [2s, (CH <sub>3</sub> ) <sub>2</sub> ] 4.53 (d, —OCH <sub>2</sub> —, J = 8) 5.40 (m, —CH=, H-3)	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	3.82 and 3.88	5.40 (m, H-3 and —CH=)	—	6.25 and 6.30 (2d, H-6, H-8, J = 2.5)
<b>1f</b>	151-152	1.75 and 1.80 [2s, (CH <sub>3</sub> ) <sub>2</sub> ] 4.53 (d, —OCH <sub>2</sub> —, J = 8)	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	3.80 and 3.90	5.40 (m, H-3 and —CH=)	—	6.25 and 6.30 (2d, H-6, H-8, J = 2.5)
<b>1g</b>	147-155	5.40 (m, —CH=, H-3) —	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	3.97	5.60 (s)	1.70 [s, (CH <sub>3</sub> ) <sub>2</sub> ] 2.67 (s, ≡CH)	7.05 (dd, H-6, J = 10 and 2.5), 7.30 (d, H-8, J = 2.5) 7.70 (d, H-5, J = 10)
<b>1h</b>	154-155	—	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	3.90 and 3.96	5.46 (s)	1.78 [s, (CH <sub>3</sub> ) <sub>2</sub> ] 2.70 (s, ≡CH)	6.47 and 6.95 (2d, H-6, H-8, J = 2.5)
<b>1i</b>	142-143	—	$\begin{matrix} \text{CH}_3 \\   \\ \text{CH}_3 \end{matrix}$	3.82 and 3.88	5.45 (s)	1.65 [s, (CH <sub>3</sub> ) <sub>2</sub> ] 2.55 (s, ≡CH)	6.55 and 7.00 (2d, H-6 and H-8, J = 2.5)

All compounds gave satisfactory C, H values; chemical shift are given in δ values (ppm) from internal TMS; coupling constants in Hz.

Table 2. Melting points and <sup>1</sup>H-NMR spectral data of compounds **2**, **3**, **4**, and **5** (in CDCl<sub>3</sub>)

Com- pound	M. P. (°C)	Me groups for <b>2</b> , <b>3</b> or Me + olefinic H for <b>4</b> , <b>5</b>	OCH <sub>3</sub> (s)	H-3 (s)	Aromatic protons
<b>2 a</b>	140-141	1.32 (d, CH <sub>3</sub> , J = 7) 1.45 and 1.55 [2 s, (CH <sub>3</sub> ) <sub>2</sub> ], 4.50 (q, =CH-, J = 7)	3.92	5.50	6.66 (d, H-6, J = 10) 7.58 (d, H-5, J = 10)
<b>2 b</b>	110-111	1.30 (d, CH <sub>3</sub> , J = 7) 1.40 and 1.50 [2 s, (CH <sub>3</sub> ) <sub>2</sub> ]	3.80 and 3.90	5.45	6.25 (s, H-6)
<b>3</b>	150-151	1.32 (d, CH <sub>3</sub> , J = 7) 1.40 and 1.50 [2 s, (CH <sub>3</sub> ) <sub>2</sub> ]	3.80 and 3.90	5.35	6.20 (s, H-8)
<b>4 a</b>	130-131	4.15 (q, =CH-, J = 7) 1.46 [s, (CH <sub>3</sub> ) <sub>2</sub> ] 5.68 (d, α-H, J = 10) 6.85 (d, β-H, J = 10)	3.95	5.55	6.70 (d, H-6, J = 10) 7.52 (d, H-5, J = 10)
<b>4 b</b>	144-145	1.42 and 1.44 [2 s, (CH <sub>3</sub> ) <sub>2</sub> ] 5.50 (d, α-H, J = 10) 6.75 (d, β-H, J = 10)	3.82 and 3.88	5.42	6.19 (s, H-6)
<b>5</b>	135-136	1.25 and 1.45 [2 s, (CH <sub>3</sub> ) <sub>2</sub> ] [2 s, (CH <sub>3</sub> ) <sub>2</sub> ] 5.55 (d, α-H, J = 10) 6.60 (d, β-H, J = 10)	3.85 and 3.90	5.45	6.35 (s, H-8)

All compounds gave satisfactory C, H values; chemical shifts are given in δ values (ppm) from internal TMS; coupling constants in Hz.

acetone or dioxan and the combined filtrate distilled. On crystallization from benzene — petroleum ether the respective prenyloxy coumarins (**1d**, **1e** or **1f**) (Table 1) were obtained.

#### *Claisen Rearrangement of Prenyloxycoumarins*

The above prenyloxycoumarins (0.5 g) in *N,N*-dimethylaniline (5 ml) were refluxed for 15 h at 200–210 °C. The solution was cooled, diluted with *Et*<sub>2</sub>O, the ether extract washed with dil. HCl (to remove dimethyl aniline) and then with sodium hydroxide solution. The remaining ether solution was dried (MgSO<sub>4</sub>), distilled and the residue crystallised from methanol to give the appropriate 4',4',5'-trimethyl dihydrofuranocoumarins (**2a**, **2b** and **3**) (Table 2).

#### *Reaction of Hydroxycoumarins in Acetone with 3-Chloro-3-methyl-but-1-yne*

The appropriate hydroxycoumarin (**1a**, **1b** or **1c**) (1 g) in dry acetone (50 ml) was refluxed for 6 h with 3-chloro-3-methyl-but-1-yne in presence of anhydrous K<sub>2</sub>CO<sub>3</sub> and potassium iodide. The solution was filtered, the residue washed with more acetone and the combined filtrate distilled. Crystallisation of the residue from benzene—petroleum ether yielded respective propargyl ethers (**1g**, **1h** or **1i**) (Table 1).

#### *Claisen Rearrangement of Propenyloxycoumarins (1g, 1h and 1i)*

The above propenyloxy coumarins (0.5 g) were refluxed with *N,N*-dimethylaniline (5 ml) for 6 h at 180–190°. Working up of the reaction mixture (as above) and crystallisation from methanol yielded the appropriate 2,2-dimethylchromopyran (**4a**, **4b**, and **5**) (Table 2).

#### *Reaction of the Hydroxycoumarins in Dioxan with 3-Chloro-3-methyl-but-1-yne*

The appropriate hydroxy coumarin (1 g) was refluxed with 3-chloro-3-methyl-but-1-yne in dioxan for 12 h in presence of anhydrous K<sub>2</sub>CO<sub>3</sub> and potassium iodide. The solution was filtered, the residue washed with more dioxan and the combined filtrate distilled. The product, on crystallisation from methanol, yielded the 2,2-dimethyl chromenopyrans (**4a**, **4b**, or **5**) (Table 2).

### Acknowledgement

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